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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.008 Å Disorder in main residue R factor = 0.045 wR factor = 0.107 Data-to-parameter ratio = 11.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title 17-crown-5 unit, $C_{24}H_{24}O_5$, comprising three benzo groups and diethylene glycol was prepared from the reaction of α, α' -dibromo-*o*-xylene and bisphenol in the presence of sodium hydride as base. This molecule offers a cavity suitable for host–guest complexes.

2,11,18,21,24-Pentaoxatetracyclo[23.4.0.0^{4,9}.0^{12,17}]-

nonaicosa-1(25),4(9),5,7,12(17),13,15,26,28-nonaene

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Comment

Since the first report on crown ethers (Pedersen, 1967), a great number of crown compounds containing one or more aromatic subunits have been investigated (Gokel & Korzeniowski, 1982; Izatt & Christensen, 1978, 1979, 1981; Vögtle & Weber, 1985; Lindoy, 1989; Weber *et al.*, 1989). A number of crown compounds bearing two macrocyclic polyether rings within the molecule, which are referred to as biscrowns, have also been synthesized (Timko *et al.*, 1974; Frensch & Vögtle, 1979; Rebek *et al.*, 1980; Lee *et al.*, 1992). In our previous paper, we reported the preparation of a new crown ether, (I), and its solid-state structure, which could be a precursor of the common-nuclear biscrown ether, bearing a benzene ring (Sim *et al.*, 2001).



We now report a new preparation of crown ether (II), the parent crown ether unit of (I), and its solid-state structure, which is the parent crown ether unit of (I). The preparation of (II) has been reported by Weber and Vögtle (1976), but the yield was rather low when KOH with dimethylformamide (DMF) was used. Therefore, alternative reaction conditions were examined so as to optimize the cyclization yield. In the reaction of α, α' -dibromo-*o*-xylene with 1,5-bis(2-hydroxyphenoxy)-3-oxapentane, the use of *n*-BuLi with tetrahydrofuran (THF) gave a 65% yield. Sodium hydride and THF provided a 72% yield but potassium *tert*-butoxide with THF gave only a moderate 50% yield. Herein, we report the optimized reaction conditions (sodium hydride and THF) for (II), as well as its solid-state structure.

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ORTEP-3 (Farrugia, 1997) diagram of (II) showing 35% probability displacement ellipsoids. C atoms are expressed with ellipsoids with only enveloping ellipses and the rest with ellipsoids with octant shading. H atoms have been omitted for clarity. The C4 atom is disordered over two positions.

Compound (II), with the atomic labelling scheme, is shown in Fig. 1. There is crystallographic mirror symmetry in (II) so that half a molecule comprises the asymmetric unit. In the diethylene glycol chain-bridging benzo groups B and B', the C10-O2-C11 [116.5 (4)°] and $C12-O3-C12^{i}$ [111.3 (5)°] angles are slightly greater than tetrahedral, while the O2- $C11-C12 [106.7 (5)^{\circ}]$ and $O3-C12-C11 [108.5 (3)^{\circ}]$ angles are very close to tetrahedral [symmetry operation: (i) 2 - x, y, z]. The dihedral angle between rings B and B' is 83.4 (1)°. In the A-to-B ring connectivity, the C3-C4-O1-C5 torsion angle is $174.9 (5)^{\circ}$ [for the minor component of the disorder associated with C4, C3-C4'-O1-C5 is $-148.6(5)^{\circ}$] which indicates that ring A is situated *trans* to ring B, with a dihedral angle of $69.2 (1)^{\circ}$ between them. In the diethylene glycol group, the O-C-C-O and two C-O-O-C torsion angles are gauche and trans, respectively, as there is one extra atom in the diethylene glycol backbone. All these conformations are very similar to those found in (I) (Sim et al., 2001). The interatomic distances $O1 \cdots O3$ of 4.580 (5) Å, $O1 \cdots O2^{i}$ of 5.344 (5) Å and $O2 \cdots O2^{i}$ of 4.941 (6) Å suggest sufficient room for the introduction of a guest atom/molecule inside the cavity. The closest intermolecular distance of 2.41 Å occurs between H11b and H12bⁱⁱ which suggests that the molecular packing is governed by van der Waals forces [symmetry operation: (ii) $x, 1 - y, \frac{1}{2} + z$].

Experimental

To a refluxing suspension of sodium hydride (15.2 mmol) in THF under nitrogen was added dropwise a solution of α, α' -dibromo-oxylene (3.79 mmol) and 1,5-bis(2-hydroxyphenoxy)-3-oxapentane (3.79 mmol) in THF over a period of 3 h. The mixture was then refluxed for an additional 24 h. After cooling to room temperature, 10% aqueous hydrochloric acid was added. The solvent was removed under reduced pressure and the residual mixture was extracted with dichloromethane. The organic layer was washed with water, dried over anhydrous magnesium sulfate, and evaporated in vacuo. The crude product was chromatographed on a silica-gel column using a mixed solvent of ethyl acetate and n-hexane (1:1) as eluent, and recrystallization from dichloromethane/*n*-hexane (1:20, v/v) gave (II) as a crystalline solid in 72% yield (m.p. 398-400 K). IR (KBr pellet) 3065, 2880, 1596, 1498, 1258, 1119, 949 and 741 cm⁻¹. ¹H NMR (CDCl₃): 87.65-7.39 (m, 4H, Ar-H), 7.10-6.88 (m, 8H, Ar-H), 5.31 (s, 4H, OCH₂Ar), 4.20 (t, 4H, OCH₂CH₂O) and 3.92 (t, 4H, OCH₂- $CH_2O).$

Crystal data

C ₂₄ H ₂₄ O ₅	Mo $K\alpha$ radiation
$M_r = 392.43$	Cell parameters from 20
Orthorhombic, $Cmc2_1$	reflections
a = 21.585 (4) Å	$\theta = 11.4 - 14.1^{\circ}$
b = 11.675 (3) Å	$\mu = 0.09 \text{ mm}^{-1}$
c = 8.092 (2) Å	T = 293 (2) K
V = 2039.3 (8) Å ³	Block, colorless
Z = 4	$0.37 \times 0.25 \times 0.20$ mm
$D_x = 1.278 \text{ Mg m}^{-3}$	

Data collection

Enraf-Nonius CAD-4 diffractometer Non-profiled $\omega/2\theta$ scans 1650 measured reflections 1597 independent reflections 708 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.011$

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.045$
$wR(F^2) = 0.107$
S = 0.98
1597 reflections
144 parameters
H-atom parameters constrained

less × 0.20 mm $\theta_{\rm max} = 30.0^\circ$ $h = 0 \rightarrow 30$ $k = 0 \rightarrow 16$ $l = -11 \rightarrow 5$ 3 standard reflections frequency: 300 min intensity decay: 1%

$w = 1/[\sigma^2(F_o^2) + (0.0360P)^2]$
+ 0.0439 <i>P</i>]
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.11 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.14 \text{ e} \text{ \AA}^{-3}$
Extinction correction: SHELXL97
Extinction coefficient: 0.0098 (9)

C4 was disordered over two positions and their sites were refined anisotropically using the PART command in SHELXL97 (Sheldrick, 1997). All H atoms were located in their idealized positions with U_{iso} constrained to be 1.2 times the equivalent isotropic displacement of the parent atoms. The number of Friedel pairs measured is 180 and the fraction of Friedel pairs measured is 0.113. As there are no heavyatom, *i.e.* Z > Si, types present, the absolute structure was not determined.

Data collection: CAD-4 EXPRESS (Enraf-Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms & Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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